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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.007 \AA$
Disorder in main residue
$R$ factor $=0.058$
$w R$ factor $=0.163$
Data-to-parameter ratio $=16.9$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## 2,7,12-Tris(4-bromobutoxy)-3,8,13-tri-methoxy-10,15-dihydro-5H-tribenzo[a,d,g]cyclononene

The title compound, $\mathrm{C}_{36} \mathrm{H}_{45} \mathrm{O}_{6} \mathrm{Br}_{3}$, adopts the rigid 'crown' conformation of the parent cyclotriveratrylene (CTV) system and possesses molecular, but not crystallographic, $C_{3}$ symmetry. The cyclotriveratrylene core is rigid.

## Comment

Cyclotriveratrylenes are excellent rigid bowl-shaped frameworks which have attracted current interest in the field of host-guest chemistry. In particular, the $C_{3}$ symmetry is attractive for the design of ligands capable of transition metal coordination, leading to complexes in which the metal could be accommodated in the cyclotriveratrylene cavity or placed in close proximity to this cavity (Thomas \& Iyenbar, 1998). Furthermore, it can be used in the field of supramolecular chemistry. In this paper, the structure of $2,7,12$-tris(4-bromo-butoxy)-3,8,13-trimethoxy-10,15-dihydro-5H-tribenzo $[a, d, g]$ cyclononene, (I), is reported (Fig. 1).

(I)

The molecule adopts the rigid 'bowl' conformation of the parent cyclotriveratrylene (CTV) system (MacNicol, 1984) and possesses molecular, but not crystallographic, $C_{3}$ symmetry. Cyclotriveratrylene and several of its derivatives form crystal structures which are not closely packed and which can thus accommodate guest molecules within voids in their structure. This property is not necessarily associated with a crown conformation. On the other hand, the existence of a


Figure 1
The molecular structure of (I), drawn with $30 \%$ probability ellipsoids. The minor occupancy disordered atoms have been omitted.


The crystal structure of (I), viewed along the $b$ axis. Dashed lines indicate hydrogen.
rigid, bowl-shaped geometry is of great importance for the formation of host-guest molecular complexes (Collet, 1987).

## Experimental

To a solution of [4-(2-bromoethoxy)-3-methoxyphenyl]methanol $(5 \mathrm{~g}, 0.019 \mathrm{~mol})$ in methanol $(25 \mathrm{ml})$ was added $70 \%$ perchloric acid $(25 \mathrm{~g}, 0.25 \mathrm{~mol})$ dropwise with chilling and stirring. The mixture was stirred overnight at room temperature. Dilution with excess water
and filtration gave a crude white solid. The solid was then chromatographed on a silica-gel column, eluting with ethyl acetate/petroleum ether ( $1 / 5$ ), to afford the pure product ( $1.1 \mathrm{~g}, 23.6 \%$ ). Colorless crystals were obtained by recrystallization from ethyl acetate.

Crystal data
$\mathrm{C}_{36} \mathrm{H}_{45} \mathrm{Br}_{3} \mathrm{O}_{6}$
$M_{r}=813.45$
Triclinic, $P \overline{1}$
$a=9.328$ (3) $\AA$ 。
$b=13.393$ (3) $\AA$
$c=15.694$ (4) A
$\alpha=76.951$ (4) ${ }^{\circ}$
$\beta=75.162(4)^{\circ}$
$\gamma=72.894(4)^{\circ}$
$V=1787.4$ (8) $\AA^{3}$

$$
\begin{aligned}
& Z=2 \\
& D_{x}=1.511 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \text { Cell parameters from } 1024 \\
& \quad \text { reflections } \\
& \theta=3.7-24.4^{\circ} \\
& \mu=3.43 \mathrm{~mm}^{-1} \\
& T=293(2) \mathrm{K} \\
& \text { Block, colorless } \\
& 0.24 \times 0.20 \times 0.16 \mathrm{~mm}
\end{aligned}
$$

Data collection
Bruker SMART CCD area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan
(SADABS; Bruker, 1997)
$T_{\text {min }}=0.432, T_{\text {max }}=0.578$
10436 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.058$
$w R\left(F^{2}\right)=0.163$
$S=1.02$
7248 reflections
428 parameters
H atoms treated by a mixture of independent and constrained refinement

7248 independent reflections 3984 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.027$
$\theta_{\text {max }}=26.4^{\circ}$
$h=-11 \rightarrow 7$
$k=-16 \rightarrow 12$
$l=-18 \rightarrow 19$

$$
\begin{aligned}
& w=1 / {\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0696 P)^{2}\right.} \\
&+1.5971 P] \\
& \text { where } P=\left(F_{o}{ }^{2}+2 F_{c}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.85 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.92 \mathrm{e} \mathrm{~A}^{-3}
\end{aligned}
$$

Atoms C33, Br3 and the H atoms on C32 and C33 are disordered, with occupancies of 0.807 (7) for $\mathrm{H} 32 A, \mathrm{H} 32 B, \mathrm{C} 33, \mathrm{H} 33 A, \mathrm{H} 33 B$ and Br 3 , and 0.195 (7) for $\mathrm{H} 33 C, \mathrm{H} 33 D, \mathrm{C} 3^{\prime}, \mathrm{H} 33 C, \mathrm{H} 33 D$ and $\mathrm{Br} 3^{\prime}$. All H atoms were positioned geometrically ( $\mathrm{C}-\mathrm{H}=0.93-0.97 \AA$ ) and refined as riding, with $U_{\mathrm{iso}}=1.2$ or 1.5 (methyl) $U_{\text {eq }}$ (parent atom).

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

## References

Bruker (1997). SADABS, SMART, SAINT and SHELXTL. Versions 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
Collet, A. (1987). Tetrahedron, 43, 5725-5759.
MacNicol, D. D. (1984). Inclusion Compounds, Vol. 2, edited by J. L. Atwood, J. E. D. Davies \& D. D. MacNicol, pp. 123-168. London: Academic Press. Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Thomas, R. M. \& Iyenbar, D. S. (1998). Indian J. Chem. Sect. B, 37, 625-627.

